

Boise State University

ScholarWorks

Materials Science and Engineering Faculty
Publications and Presentations

Micron School for Materials Science and
Engineering

10-13-2020

Mechanochemical Conversion Kinetics of Red to Black Phosphorus and Scaling Parameters for High Volume Synthesis

Samuel V. Pedersen
Boise State University

Florent Muramutsa
Boise State University

Joshua D. Wood
Promethean Consulting, LLC

Chad Husko
Iris Light Technologies, Inc.

David Estrada
Boise State University

See next page for additional authors

Publication Information

Pedersen, Samuel V.; Muramutsa, Florent; Wood, Joshua D.; Husko, Chad; Estrada, David; and Jaques, Brian J. (2020). "Mechanochemical Conversion Kinetics of Red to Black Phosphorus and Scaling Parameters for High Volume Synthesis". *NPJ: 2D Materials and Applications*, 4, 36. <https://doi.org/10.1038/s41699-020-00170-4>

Authors

Samuel V. Pedersen, Florent Muramutsa, Joshua D. Wood, Chad Husko, David Estrada, and Brian J. Jaques

ARTICLE OPEN



Mechanochemical conversion kinetics of red to black phosphorus and scaling parameters for high volume synthesis

Samuel V. Pedersen^{1,2}, Florent Muramutsa¹, Joshua D. Wood³, Chad Husko⁴, David Estrada^{1,2,5} and Brian J. Jaques^{1,2}✉

Adopting black phosphorus (BP) as a material in electronic and optoelectronic device manufacturing requires the development and understanding of a large-scale synthesis technique. To that end, high-energy planetary ball milling is demonstrated as a scalable synthesis route, and the mechanisms and conversion kinetics of the BP phase transformation are investigated. During the milling process, media collisions rapidly compress amorphous red phosphorus (RP) into crystalline, orthorhombic BP flakes, resulting in a conversion yield of $\approx 90\%$ for ≈ 5 g of bulk BP powder. Milling conversion kinetics, monitored via ex situ x-ray diffraction, manifest a sigmoidal behavior best described by the Avrami rate model with each impact of sufficient energy (>25 mJ) producing BP nuclei; the process appears to be limited by grain growth. Using a kinematic model for ball trajectories and impact energies, the optimum milling condition is determined to be an impact energy near ≈ 25 mJ and a milling dose near ≈ 100 kJ/gram. Photoexcitation of exfoliated BP flakes reveals emission in the near-infrared, indicating the formation of few-layer BP, a promising advance for optoelectronic device applications.

npj 2D Materials and Applications (2020)4:36; <https://doi.org/10.1038/s41699-020-00170-4>

INTRODUCTION

The surge of research into two-dimensional materials after the discovery of graphene in 2004 has led to investigating the monolayer limit of layered materials that include transition metal dichalcogenides (TMD), hexagonal boron nitride (hBN), MXenes, layered elemental metals (Te, Sn, Ge, As, Se)¹, black phosphorus (BP), and beyond. Orthorhombic BP, the high pressure and temperature phase of elemental phosphorus, consists of a puckered hexagonal sheet of covalently bonded atoms. In bulk form, the individual sheets are bound together by van der Waal forces; however, new properties emerge due to quantum size effects when exfoliated into the few-layer regime. A property of note is BP's direct bandgap, which changes from 0.3 eV in the bulk to ≈ 1.7 eV in the monolayer limit^{2,3}. Additionally, BP shows promise in optoelectronic device applications⁴⁻⁶ from its direct bandgap as BP emissions are between 700 and >4000 nm, which is the near- to mid-infrared wavelength range.

Nevertheless, a standing challenge for BP is its synthesis. Traditionally, generating BP necessitates either high pressure and elevated temperature as in the diamond anvil method first pioneered by Bridgman⁷ to phase transform either white or red phosphorus (RP), or the use of a low-pressure mineralizer assisted chemical vapor transport (CVT)⁸. While CVT is the established commercial production process, it suffers from high cost and low volume. In the CVT method, RP, tin iodide, and tin are sealed inside a glass ampule that is evacuated to $\approx 10^{-3}$ Torr, heated above 825 K, and allowed to cool slowly yielding BP crystals over a 1–3 day period that grow outward from an intermediate Sn-P-I compound⁹.

Alternatively, high-energy planetary ball milling (HEPBM) constrains the pressure induced phase transformation inside a steel vessel under an inert atmosphere, alleviating potential safety hazards associated with the CVT method due to over-pressurization of the glass ampules or incorrect cooling rates.

Additionally, no expensive and toxic mineralizers (i.e., tin, iodine) are required. In order to facilitate the scale-up of BP synthesis via HEPBM, details such as the conversion rate as a function of impact energy (i.e., ball density and size) and the cumulative energy input per gram of product formed are required. Several reports¹⁰⁻¹³ of HEPBM on BP synthesis have explored the processing effects on synthesis time such as ball to powder mass ratio and rotation speed of the milling vessel but have not reported the scaling parameters. HEPBM is readily transferred to industrial mills capable of kilogram production capacities, provided that the energetics and kinetics of the phase transformation are well understood.

This study determines the necessary impact energy, the apparent rate constants, and the specific milling dose of the RP to BP phase transformation via HEPBM. By using closed-form analytical models of the kinematic motion inside the mill¹⁴ along with estimates for the phase fraction evolution, an empirical relationship is established showing that higher impact energies lead directly to higher conversion rates while the total energy dose is shown to be an independent constant. Furthermore, the low cost nano-crystalline BP upon exfoliation is shown to be suitable for infrared optoelectronic applications, thereby paving the way for commercial scale-up opportunities.

RESULTS

Black phosphorus characterization

Visually, the powder changes color from a burnt umber red to a dull black with various shades of red/brown/black with increasing BP content, as shown in Fig. 1a. The bulk red and black powders exhibit similar morphology and agglomerations. The X-ray diffraction (XRD) patterns for the conversion of micron sized amorphous RP powder into orthorhombic BP are shown in Fig. 1b as a function of milling duration at the highest experimental

¹Micron School of Materials Science and Engineering, Boise State University, Boise, ID 83725, USA. ²Center for Advanced Energy Studies, Idaho Falls, ID 83401, USA. ³Promethean Consulting, LLC, Oak Park, IL 60302, USA. ⁴Iris Light Technologies, Inc., Chicago, IL 60622, USA. ⁵Idaho National Laboratory, Idaho Falls, ID 83402, USA. ✉email: brianjaques@boisestate.edu

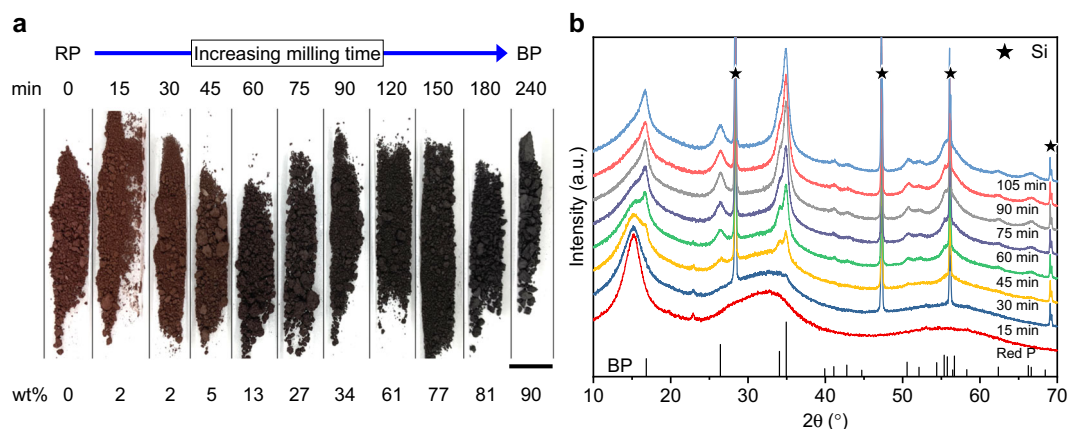


Fig. 1 Characterization of the mechanochemical conversion of RP to BP. **a** Optical image of bulk powders revealing the color progression that occurs during increased milling time via high-energy ball milling; the transition from red to brown to black is a direct result of the mixture of amorphous RP and crystalline BP present (scale bar is 1 cm). **b** Powder x-ray diffraction (XRD) patterns of RP during conversion to BP showing an increase in the intensity of the main BP (040) and (111) peaks with increasing milling time; here, all patterns vertically offset correspond to runs at a constant speed 600 rpm, black bars are BP (PDF 03-065-2491), and silicon standard peaks are marked.

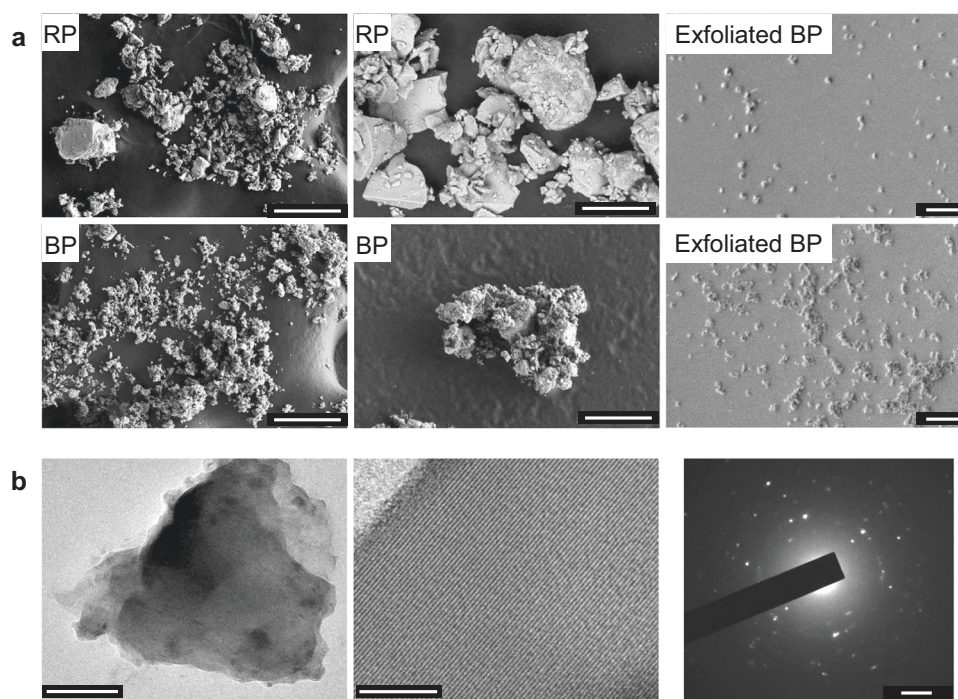


Fig. 2 Electron microscopy shows particle size reduction through BP synthesis and exfoliation while retaining polycrystallinity. **a** Scanning electron microscopy (SEM) images presenting faceted morphology of large particle sizes for RP, smaller particle sizes for BP, and narrow particle size for exfoliated BP flakes that consist of crystalline domains embedded in a larger bulk amorphous particle with plate-like morphology (SEM scale bars from left to right are 100 μm , 10 μm , and 2 μm , respectively). **b** Transmission electron microscopy (TEM) images and a selected area electron diffraction (SAED) pattern of polycrystalline BP manifesting large lattice domains and converted BP crystallinity. TEM scale bars from left to right are 50 nm, 10 nm, and 2 nm^{-1} respectively.

milling intensity evaluated in this study. The as-received RP exhibits three broad peaks, with the first defined diffraction peak at $\approx 17^\circ 2\theta$. The powder diffraction file (PDF 03-065-2491) from the Inorganic Crystal Structures Database (ICSD¹⁵) for BP is also shown, with relative peak intensities and positions indicated with the vertical line markers. A gradual shift to higher angles of the first sharp diffraction peak of RP occurs. By contrast, the broader high angle peaks generally decrease in intensity and width as the crystalline BP peaks emerge. XRD scans versus milling rpm using an internal silicon standard and the associated calibration curve are included in the Supplementary Figs 1–5.

The scanning electron microscopy (SEM) images of Fig. 2a show that the RP starting powder exhibits an irregular faceted morphology with particle sizes from ~ 0.1 to 100 μm . The converted BP exhibits similar faceted morphology but has a smaller particle size distribution below 10 μm . The powders appear to consist of soft aggregates of BP crystals embedded into amorphous particles which break down into flakes consisting of crystalline domains surrounded by amorphous matrix under mild ultrasonication. To further verify the crystalline BP domains within the amorphous flakes, transmission electron microscopy (TEM) images and selected area electron diffraction (SAED)

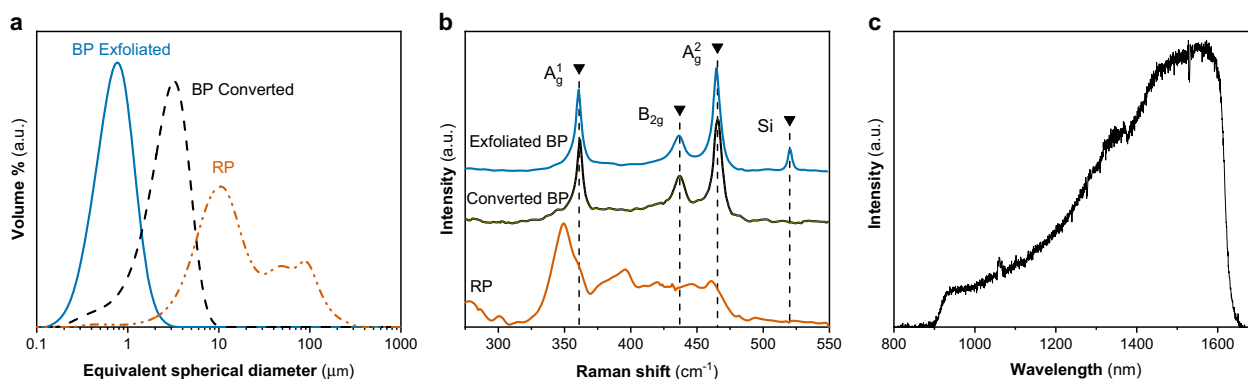


Fig. 3 Characterization of mechanochemically synthesized and exfoliated few-layer BP. **a** Particle size analysis of RP, as-converted BP, and ball mill exfoliated BP showing progressively narrower and smaller size distributions as a result of conversion, comminution, and exfoliation. **b** Raman spectra confirming the amorphous nature of RP, as well as the crystalline nature of both converted and exfoliated BP. **c** Photoluminescence (PL) emission of exfoliated BP on SiO₂ with a broad spectrum due to the bilayer to multilayer BP in the exfoliated material.

patterns are taken and presented in Fig. 2b, revealing the presence of crystalline BP domains embedded within the amorphous flakes and agglomerated powders. The volumetric particle size distributions obtained from solution-based laser diffraction particle size analysis (PSA) shown in Fig. 3a shows non-uniform size distribution for RP. This is an expected result considering the large variation in particle sizes observed in the SEM images (see Fig. 2) and the high sensitivity of volumetric PSA for detection of few but large particles. By contrast, the milling comminution process results in a nearly single mode distribution of the converted BP material, while exfoliation breaks up soft BP aggregates and produces flakes consisting of embedded crystal domains within an amorphous matrix.

To complement the XRD and TEM results on the converted material and verify the crystallinity of the exfoliated BP, Raman spectra are collected on the as-received RP, synthesized BP powder, and exfoliated BP flakes as shown in Fig. 3b. The expected BP vibrational modes³ A_{1g}, B_{2g}, and A_{2g}, respectively, are clearly resolved, while the peak at 520 cm⁻¹ is from the silicon substrate. The slightly raised background between 350 to 475 cm⁻¹ is likely attributed to residual RP¹¹.

As aforementioned, the direct bandgap of BP that spans 0.3–1.7 eV makes BP suitable for near-infrared wavelength applications. Few-layer (ca. 5 layers) BP exhibits optically pumped emission^{2,5,16} near ≈1575 nm, whereas tri- and bi-layer BP emit at shorter wavelengths. The photoluminescence (PL) spectra for ball mill converted BP after exfoliation is shown in Fig. 3c. The broad emission from 950 to 1600 nm indicates the flakes contain bi-layer, tri-layer, and multilayer BP crystalline domains; however, further efforts to fully separate the various domains by lateral size or layer thickness by such means as density gradient centrifugation must be demonstrated prior to use in photonic applications^{2,16,17}.

Conversion kinetics

Detailed individual reaction steps during mechanical milling processes are challenging to elucidate, although recent progress with synchrotron X-ray and Raman spectroscopy is now capable of providing in situ monitoring of reaction steps¹⁸. Nevertheless, empirical models based on ex situ measurements can provide general insight into the overall kinetic behavior, such as determining if the process accelerates to completion, undergoes deceleration, or progresses at a constant rate^{19–21}. For evaluating the kinetics of a process, typically through a series of experiments, the energy input (such as a constant temperature or constant heating rate) is varied systematically and the evolution of the phase fraction, α , as a function of time is used to extract the rate constant, k ^{19,22}.

In mechanochemistry, the two kinetic models often used for solid-state phase transformations are the classical

Johnson–Mehl–Avrami–Kolmogorov (JMAK)^{23–25}, and a deterministic collision driven model derived by Delogu and Cocco (DC)^{20,26,27}. Both models have been used to obtain the rate constants for single-phase systems undergoing transformations or amorphization and have been applied to multiphase systems undergoing mechanical alloying^{23–25}. After obtaining the apparent rate constants as a function of milling parameters, the relationship between the milling impact energy, E_b , and the rate constants can be determined. If a single impact is sufficient to drive the conversion to completion, then a linear dependence of the rate constants and the impact energies is expected²⁸.

Simulations and in situ monitoring^{27,29–36} of media interactions in high-energy ball mills have shown agreement with the underlying assumptions and estimated impact energies obtained through kinematic models of high-energy planetary ball mills. While existing models differ in determining the media velocity at impact^{14,30,31,34,37}, all models assume the kinetic energy imparted to the media through collisions with the vial walls is transferred into the powder during impact, at least to a first order approximation (some losses due to friction, heat generation, and inelasticity are expected). Nonetheless, the impact energy, E_b , is simply related to the media mass, m_b , and the impact velocity, v_{in} , as:

$$E_b = \frac{1}{2} m_b v_{in}^2 \quad (1)$$

By extension, the cumulative dose, E_{dose} , is a function of the number of balls, N_b , the collision frequency, f_b , the milling time, t , and powder mass, m_p , as:

$$E_{dose} = \frac{E_b f_b N_b t}{m_p} \quad (2)$$

Using the conversion data obtained from XRD, the method described by Galwey et al.¹⁹ was employed to generate a master plot to determine which subset of solid-state reaction models are applicable, as shown in Supplementary Fig. 6. The JMAK model closely matches the dataset, while the DC model follows the data above 30% conversion. As seen in Fig. 4a for 300 rpm, the rate of conversion slows down, and this transition is visually distinct for the 300 rpm run versus the higher intensities of 400–600 rpm. Similarly, for 200 rpm, the conversion occurs slowly with an induction time of nearly 100 h before BP is observed. Transforming the data using the JMAK model, Fig. 4b, or the DC model, Fig. 4c, shows two or three linear segments. In practice, line segments that show different slopes are treated individually with linear regression lines, as it has been shown that these distinct transitions are often due to changes in growth dimensionality³⁸.

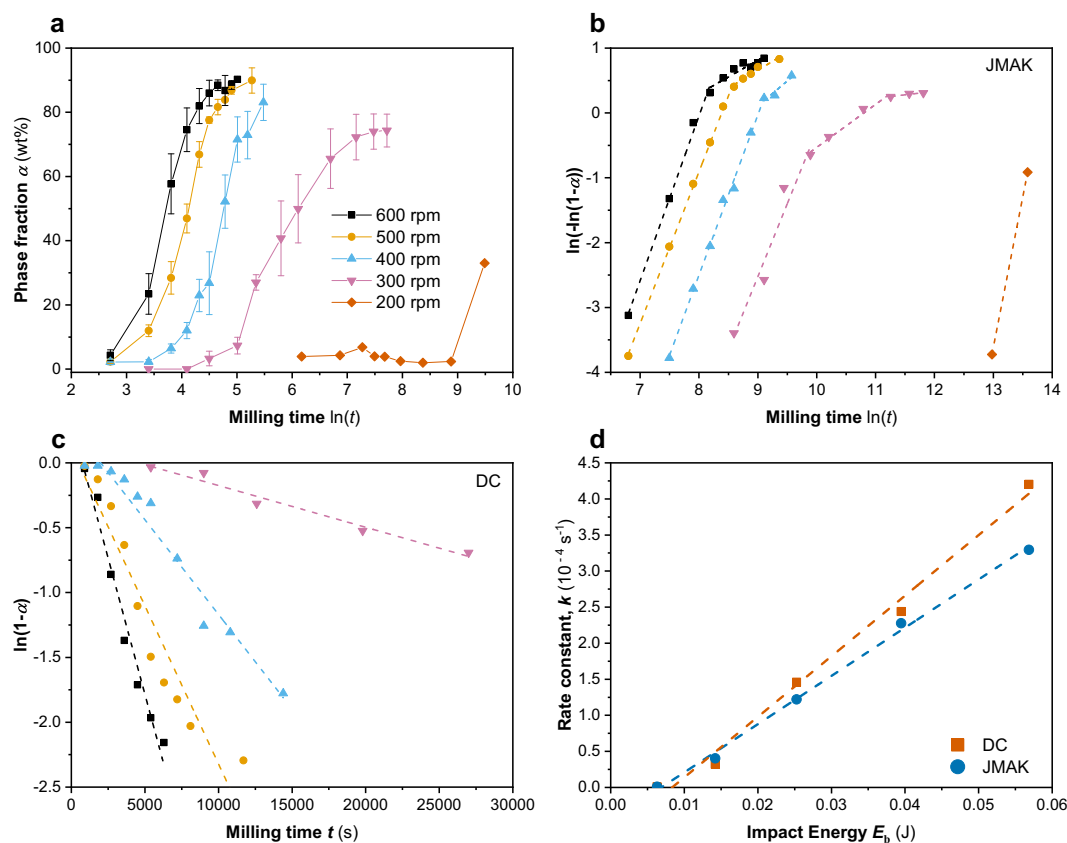


Fig. 4 Mechanochemically induced phase transformation kinetics of RP to BP using the JMAK and DC models. **a** XRD-based phase fraction versus logarithmic time showing representative S-curves for each rpm condition; higher rpm results in faster conversion while a dramatic reduction occurs at and below 300 rpm. An S-shape curve is typical of nucleation and growth dominated solid-state processes. Error bars represent one standard deviation, lines are visual aids. **b** JMAK linearization plots for extracting rate constants and exponents, lines are linear regression fits for each straight section; the slopes remain relatively constant indicating the conversion mechanism remains constant, while the kinks at higher phase fraction indicate a possible mechanism change. **c** DC model linearization plot excluding the 200 rpm data, lines are linear regression fits; the slopes become steeper (higher rate constant) with increasing rpm. **d** Linear regression fits for rate constants versus calculated impact energies; both the JMAK and DC models show the linear dependence of the rate constant on the impact energy confirming the assumption that a single impact is sufficient for converting the entrapped RP powder between media collisions to BP.

As a reminder, the DC model is, mathematically, a restricted form of the JMAK model that fixes the exponent $n = 1$, resulting in an apparent S-shape about the linear regression line in the DC linearization plot. Functionally, this restriction prevents making inferences about the nucleation and growth behavior afforded by the JMAK model, but as shown in Fig. 4d the extracted rate constants are still comparable. In all cases, both the JMAK and DC models appear to be valid, and the linear fits model the data up to 90% conversion well with values for the coefficient of determination (a goodness of fit statistic relating the percentage of the predicted response variation that is modeled by the regression line), R^2 , above >95%. Deviations¹⁹ from the model are likely due to the initial estimation of the phase fraction below 10 wt% and above 70 wt%. In these regions, there appears to be a consistent decrease in conversion rate for all milling intensities, implying a mechanistic change that requires higher mechanical impact energy to progress the conversion. Supplementary Figure 7 (plotted on a log scale) includes the 200 rpm curve; the 200 rpm dataset is not shown in Fig. 4c, as the induction time obscures the higher rpm behaviors. Figure 4d shows the linear dependence of the apparent rate constants as a function of the calculated impact energy. The calculated impact energies and milling intensities using the kinematic model as applied to the HEPBM and the ascertained kinetic parameters for both the JMAK and DC models are presented in the Supplementary Table 1 and Table 2, respectively.

Figure 5 shows the phase fraction evolution as a function of the total energy imparted to the powder on a mass basis, revealing three distinct behaviors. First, for higher milling intensities (i.e., mechanical work performed) above 300 rpm, the total dose required for complete conversion is the same. Second, below 300 rpm, there is a transition at low phase fractions (about 30 wt %) requiring progressively longer milling times, as well as a plateau near 75 wt%. This plateau indicates a potentially limiting step which cannot be overcome by the impact energy at 300 rpm. Third and finally, at even lower energy (i.e., 200 rpm), nucleation of BP is evident, but only after a long induction time of 200+ hours.

DISCUSSION

The reduction in particle size of RP as it converts to BP due to comminution is typical of ball milling and is desirable as it increases the surface area available for reactions and promotes a single mode particle size distribution which benefits downstream processes such as exfoliation. The broad BP XRD peaks in Fig. 1b indicate a high degree of nano-crystallinity while the broad base of the peak at $17^\circ 2\theta$ indicates some amorphous material is still present. Supplementary Figures 8 and 9 show TEM images of randomly orientated lattice planes surrounded by an amorphous matrix within a single exfoliated flake, indicative of the nature of point impacts that have occurred and that presumably each impact will produce a local formation of BP within the bulk RP

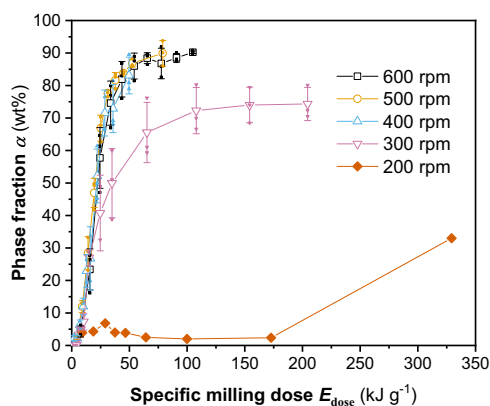


Fig. 5 Extent of conversion for RP to BP as a function of specific milling dose. Milling at 400 rpm or higher results in an equivalent energy dose to obtain the same extent of conversion. Milling at <400 rpm (i.e., 300 and 200 rpm) results in incomplete conversion due to an apparent energy threshold. Phase fraction data normalized by milling dose estimates. Error bars represent one standard deviation.

particle^{10,12}. Amorphous RP possesses some degree of short and medium range order; it has been described as being a 3, 4, and 5 sided cage-like structure linked together by phosphorus atoms in a polymer like manner^{39,40}. Presumably, the phosphorus cages are collapsed into the buckled planes due to the high localized compressive forces from the media collisions^{39,41,42}. The Raman spectra in Fig. 3b on HEPBM synthesized powder (exfoliated in IPA) shows only modes that correspond to BP, which indicates no significant RP is present. Together with the TEM data, the lack of features corresponding to RP indicates that amorphous BP is likely present^{43–46}. High pressure conversion of RP thin films into BP has also reported the formation of amorphous BP⁴⁴. High-energy ball milling has been used to produce amorphous metals and even cycle between crystalline and amorphous phases of alloys by controlling the milling intensity and duration⁴⁷. In light of this, the coexistence of both crystalline and amorphous, highly disordered, BP within ball mill synthesized BP flakes is unsurprising and has been observed through nuclear magnetic resonance spectroscopy¹¹.

While some fraction of BP is present in the XRD patterns at 200 and 300 rpm, such conversion rates are economically impractical due to the power consumption; namely, mill motors constitute the highest recurring electrical costs. The 200-rpm milling intensity shows an entirely different behavior which may be due to the tendency of media to slide along the walls of the vessel with few high-energy collisions at low vial rotation speeds. As a result of the less energetic impacts and low frequency of collision, a significantly higher milling dose is required to producing meaningful quantities of BP at 200 rpm.

From Fig. 4d, the x -axis intercept for a negligible rate constant, $k = 0$, ideally indicates a minimum impact energy, E_b , near 0.5 mJ, but these values do not reflect the apparent change above 70 wt%. As such, this minimum value most likely represents the energy required to form stable nuclei but with no practical grain growth. The utility of Fig. 4d is that extrapolation of an expected rate constant due to higher impact energies is feasible, up to a limit set by the media size, density, vessel diameter, and achievable rpm.

The main scaling parameter for HEPBM is obtained from Fig. 5 and this dictates the optimum milling condition. The product of the milling intensity, $E_b \cdot f_b \cdot N_b$, and time, t , is the same for milling at 600, 500, and 400 rpm. That is, above a certain impact energy ($E_b = 25$ mJ), the conversion takes the same total energy input. The trade-off then is the milling time versus potential media wear and design considerations for the mill motor. An apparent specific

milling dose of ≈ 100 kJ/gram is required to achieve a BP conversion above 90 wt%. While a higher milling dose may improve the yield, it may also introduce more defects into the crystalline domains, as well as produce more media wear and possible contaminations from the milling vessel and media.

The results presented here show that the conversion of RP to BP follows the characteristic S-shape of sigmoidal processes indicative of nucleation and growth dominated phase transformation. The rate constants obtained by applying the JMAK and DC models in Fig. 4 show the expected linear behavior as a function of milling intensity; accordingly, higher impact energies and higher collision frequency should result in more frequent nucleation events and, hence, a faster conversion rate. One benefit of the DC model is that it is less sensitive to induction periods than the JMAK model, such that the extracted rate constants better represent the overall conversion rate. However, the JMAK model allows for inferences into the nucleation and growth behavior.

Using the JMAK model, the values of the exponent can potentially provide insight into the dynamics of nucleation and crystal growth. The calculated exponent value is $n \approx 2.3$ – 2.5 , suggesting that RP to BP undergoes continuous nucleation and subsequent interfaced controlled transformation in two-dimensions^{48,49}. The reduced exponent value below 3 can be attributed to the anisotropic growth and blocking effects, or it could be an outcome of a mixture of 3D and 2D growth with site saturation^{48–51}. In theory, polymorphic phase transformations are categorized as either displacive (rapid and diffusionless) or reconstructive (slow and diffusion limited)^{52–54}. In practice, assumptions regarding diffusion, preferred nucleation sites such as defects, grain boundaries, or dislocations, and experimentally determined crystal growth habit (i.e., spherical, disc, or rod growth) are required to make reliable inferences from the JMAK exponent.

However, the unique amorphous to crystalline transformation of RP to BP does not fit cleanly into any phase transformation category⁵³. We argue that RP to BP is likely a rapid diffusionless, displacive transformation, provided that RP bonds are broken either through thermal or strain input. Generally, high applied pressure inhibits diffusion due to the increase in repulsive forces upon shortening the interatomic bond distance⁵² but can enable rapid short displacements via re-orientation⁵². By contrast, grain growth itself is typically a thermally driven process. High pressure studies on the white phosphorus (≈ 1.2 GPa) or RP (≈ 8 GPa) to BP transition occurred almost instantly after an incubation time of 10–15 min when exceeding a minimum temperature of 200°C ^{7,41,43,55–58}. This supports the notion of rapid interface propagation instead of a time-dependent diffusion-controlled grain growth. Comparing our mechanochemical results to Xiang et al.⁴⁴ and to those obtained by Bridgman⁵⁹ using rotating diamond anvils (≈ 4 GPa), we contend that the plastic strain facilitated by the direct compression of RP between colliding media breaks molecular bonds within the RP polymer chains, leading to the rapid formation of the orthorhombic crystalline phase. The TEM results here of flakes show clear nanometer-sized lattice domains randomly orientated surrounded by regions of an amorphous matrix, consistent with random nucleation but limited grain growth. Similar behavior was observed with static anvil experiments at a higher pressure (≈ 6.8 GPa) but a much lower temperature ($\approx 25^\circ\text{C}$) than typical synthesis conditions. Xiang et al.⁴⁴ revealed a slow transformation rate of tens of hours in which the rhombohedral (A7) structure of metallic BP nucleates first from RP and persists indefinitely (up to 71 h). However, the material rapidly converts to orthorhombic BP upon decompression⁴⁴. Collisions in HEPBM are effectively microforges where the RP is entrapped between the media in a millisecond timescale collision. Upon the media rebounding, the BP crystallizes⁴⁷.

In support of our hypothesis, the fact that the rhombohedral crystal structure of BP forms primarily under the applied

hydrostatic pressure without external heating indicates the possibility of bulk nucleation of a two-dimensional highly disordered intermediate⁴¹. Nonetheless, the absence of heat input or strain will produce sluggish nucleation, as evidenced by the in situ time-resolved Raman spectra of Xiang et al.⁴⁴. Their high-resolution TEM data on quenched samples held under static loading for ≈ 36 h revealed numerous nano-crystalline domains of orthorhombic BP surrounded by amorphous phosphorus, similar to the TEM results here obtained on mechanochemically synthesized BP. Supplementary Figure 10 shows the in situ temperature measurements obtained during mechanochemical conversion; while the global temperature does not exceed 40 °C, it is possible that the local impact energy could heat the powder substantially^{60–63}. Still, the temperature rise is also highly dependent upon the material undergoing the treatment (i.e., if the process results in an exothermic heat release such as in the milling of thermite, metal oxide formation, or sulfide reactions)^{20,47,64}. We exclude the possibility of high localized temperatures at the point of impacts, as we do not observe substantial grain sizes. Furthermore, recent work⁶⁵ comparing torsional or rotating diamond anvil methods to mechanochemistry shows striking similarities in that the generated plastic strain can reduce the pressure/temperature requirements for phase transformations⁶⁶ in Si, Ge, boron nitride^{67,68}, and zirconium⁶⁹ with similar nanostructures.

In practice, knowing the rate constant and specific milling dose enables determination of the mill design (diameter, motor horsepower, media size, and quantity) in order to achieve a given production rate. The ascertained specific milling dose of ≈ 100 kJ/gram and corresponding minimum impact energy near 25 mJ/impact provide meaningful parameters for scale-up either by number of milling stations (i.e., 100 g capable planetary ball mills) or by switching to larger tumbler mills (i.e., 1 kg capacity) and running for the appropriate milling time derived by considering the rate constant, impact energy, and desired amount of product; the difference between the planetary mill versus the tumbler mill being the rate of energy deposition versus the synthesis time⁷⁰.

In conclusion, we have ascertained the rate dependency on impact energy and the scaling parameter (milling dose near 100 kJ/gram) required to convert low cost RP into BP by the highly scalable high-energy ball milling method. While the mechanical impacts readily form BP nuclei, the overall process is grain growth limited; further efforts should seek to enhance the grain growth. The exfoliated BP flakes containing embedded crystalline domains within an amorphous matrix produces PL spectra with emission in the near-infrared range, thereby revealing the potential use of this synthesis route provided improved exfoliation methods are developed to isolate layer-dependent BP nanocrystals. Producing commodity-grade BP is an enabling step to employ the nanomaterial in commercial electronic and optoelectronic devices.

METHODS

In the milling experiment, elemental red phosphorus (1–5 g; Alfa Aesar, 97% purity, -60 mesh) and stainless steel media (10 mm diameter, 100 g) were weighed and loaded into 250 mL stainless steel vessels inside an argon atmosphere glovebox ($P_{O_2} < 0.1$ ppm) and sealed with internal gas pressures between 90 and 95 kPa at room temperature. The ball to powder mass ratio was 20:1. All milling experiments were conducted with the PM100 planetary ball mill (Retsch, Germany) with pauses in the milling cycles (30 min on, 30 min off) to prevent overheating of the unit and to switch the direction of flywheel rotation. Milling conditions (rpm and duration) were varied between 200 and 600 rpm and 0.15 to 25 h, depending upon the intent of the studies. For the kinetic study, each run was completed three times except the 200 rpm run which was allowed to run once for 220 h. The three runs at 300, 400, 500, and 600 rpm were paused periodically to obtain samples for phase fraction analysis. For each sampling interval, the sealed vessel was first transferred into an argon glovebox. An appropriate amount of powder was extracted from the free

powder at the bottom of the milling vessel and sealed inside glass vials; no powder was added back into the milling vessel. The vessel was resealed and transferred out of the glovebox to resume the milling run. An initial milling run at 600 rpm for 2 h with 1 gram of red phosphorus was used to condition the media and vessel surfaces before each milling study. The milling vessel was wiped clean between each study but was not cleaned with isopropyl alcohol, thereby preserving the conditioned state of the vessel.

To obtain few-layer BP suspensions, the as-converted BP powders (0.25 g) are solvent exfoliated in isopropanol (30 mL) using a low-energy mixer mill (MTI, SFM-2) with (1 mm) diameter chrome steel ball media sealed inside a 50 mL plastic container for 24 h at a rotational speed of 150 rpm. The suspension is separated from the media using a sieve (-300 mesh) and stored in glass vials. The suspension is drop cast onto silicon substrates for Raman and PL measurements.

Powder XRD data was obtained in the Bragg-Brentano geometry (CuK α , Rigaku MiniFlex 600, Japan) on 0.165 g of powder. The internal standard method using NIST Silicon 640D powder (ICSD PDF 00-027-1402) was used to generate a calibration curve for estimates of the phase fraction obtained.

Powder Raman spectroscopy (Raman) data was obtained (Horiba LabRam HR Evolution, Japan) on densely packed samples (0.01 grams) prepared inside an argon glovebox that were sealed inside a custom-built enclosure with a sapphire window to prevent oxidation and ignition during measurements (both red and black phosphorus in finely divided powder form can be readily ignited in air even when using neutral density filters and low laser power settings). A 50-mW 532-nm laser source (Nd:YAG) was used to generate spectra collected with a Si CCD. Spectra obtained at $\times 20$ objective with double acquisitions for 60 s at ten random locations throughout the bulk surface using the 1800 lines/mm grating.

PL measurements were obtained using a Princeton Aston spectrometer with a NIRvana InGaAs thermoelectrically cooled array and an excitation wavelength of 532 nm at 10 mW. Data were collected for 5 s using a 100X near-infrared objective.

Solution-based laser diffraction PSA was performed on powders ultrasonically dispersed in isopropanol for 1 min and cycled through the Horiba LA-950 particle size analyzer and solution flow cell; volumetric distributions were obtained using dual 405 and 650 nm laser and light diodes. The complex indices of refractions for RP⁷¹ and BP⁷² were taken from literature.

SEM was performed on powders pressed onto carbon tape and imaged at low beam currents and accelerating voltages using a FEI Teneo FESEM. Exfoliated BP powders were drop cast onto cut SiO₂ wafers and imaged under similar conditions using secondary and backscattered electron imaging modes.

TEM was performed using a JEOL-JEM-2100, Japan, with Orius SC1000 CCD camera; exfoliated samples were drop cast onto lacy carbon grids and transferred into the vacuum chamber within 5 min. A low accelerating voltage of 100 keV was used with minimal dwell times roughly 30 s for imaging to prevent the amorphization of the flakes.

DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author upon reasonable request.

CODE AVAILABILITY

All relevant code is available from the authors on request.

Received: 1 June 2020; Accepted: 19 September 2020;

Published online: 13 October 2020

REFERENCES

1. Pumera, M. & Sofer, Z. 2D Monoelemental arsenene, antimonene, and bismuthene: beyond black phosphorus. *Adv. Mater.* **29**, 1605299 (2017).
2. Zhang, S. et al. Extraordinary photoluminescence and strong temperature/angle-dependent Raman responses in few-layer phosphorene. *ASC Nano* **8**, 9590–9593 (2014).
3. Kang, J. et al. Solvent exfoliation of electronic-grade, two-dimensional black phosphorus. *ACS Nano* **9**, 3596–3604 (2015).
4. Peng, R. et al. Midinfrared electro-optic modulation in few-layer black phosphorus. *Nano Lett.* **17**, 6315–6320 (2017).

5. Husko, C. et al. Silicon-phosphorene nanocavity-enhanced optical emission at telecommunications wavelengths. *Nano Lett.* **18**, 6515–6520 (2018).
6. Jin, X. et al. 102 fs pulse generation from a long-term stable, inkjet-printed black phosphorus-mode-locked fiber laser. *Opt. Express* **26**, 12506–12513 (2018).
7. Bridgman, P. W. Two new modifications of phosphorus. *J. Am. Chem. Soc.* **36**, 1344–1363 (1914).
8. Lange, S., Schmidt, P. & Nilges, T. Au₃SnP₇@Black phosphorus: an easy access to black phosphorus. *Inorg. Chem.* **46**, 4028–4035 (2007).
9. Wang, D. et al. Revisiting the growth of black phosphorus in Sn-I assisted reactions. *Front. Chem.* **7**, 21 (2019).
10. Zhou, F. et al. Growth mechanism of black phosphorus synthesized by different ball milling techniques. *J. Alloy Compd* **784**, 339–346 (2019).
11. Ferrara, C. et al. Efficiency and quality issues in the production of black phosphorus by mechanochemical synthesis: a multi-technique approach. *ACS Appl. Energy Mater.* **2**, 2794–2802 (2019).
12. Park, C. M. & Sohn, H. J. Black phosphorus and its composite for lithium rechargeable batteries. *Adv. Mater.* **19**, 2465–2468 (2007).
13. Sun, C. et al. One-pot solventless preparation of PEGylated black phosphorus nanoparticles for photoacoustic imaging and photothermal therapy of cancer. *Biomaterials* **91**, 81–89 (2016).
14. Kakuk, G., Zsoldos, I., Csanady, A. & Oldal, I. Contributions to the modelling of the milling process in a planetary ball mill. *Rev. Adv. Mater. Sci.* **22**, 21–38 (2009).
15. Bergerhoff, G. & Brown, I. D. *Crystallographic Databases*. 77–95 (International Union of Crystallography, Chester, 1987).
16. Kang, J. et al. Stable aqueous dispersions of optically and electronically active phosphorene. *Proc. Natl. Acad. Sci. USA* **113**, 11688–11693 (2016).
17. Kang, J., Sangwan, V. K., Wood, J. D. & Hersam, M. C. Solution-based processing of monodisperse two-dimensional nanomaterials. *Acc. Chem. Res.* **50**, 943–951 (2017).
18. Užarevič, K., Halasz, I. & Friščić, T. Real-time and in situ monitoring of mechanochemical reactions: a new playground for all chemists. *J. Phys. Chem. Lett.* **6**, 4129–4140 (2015).
19. Galwey, A. K. & Brown, M. E. Isothermal kinetic analysis of solid-state reactions using plots of rate against derivative function of the rate equation. *Thermochim. Acta* **269–270**, 1–25 (1995).
20. Humphry-Baker, S. A., Garroni, S., Delogu, F. & Schuh, C. A. Melt-driven mechanochemical phase transformations in moderately exothermic powder mixtures. *Nat. Mater.* **15**, 1280–1286 (2016).
21. Colacino, E. et al. Processing and investigation methods in mechanochemical kinetics. *ACS Omega* **3**, 9196–9209 (2018).
22. Avramov, I. & Šesták, J. Generalized kinetics of overall phase transition explicit to crystallization. *J. Therm. Anal. Calorim.* **118**, 1715–1720 (2014).
23. Blázquez, J. S., Manchón-Gordón, A. F., Ipus, J. J., Conde, C. F. & Conde, A. On the use of JMAK theory to describe mechanical amorphization: a comparison between experiments, numerical solutions and simulations. *Metals* **8**, 450 (2018).
24. Matsuoka, M., Hirata, J. & Yoshizawa, S. Kinetics of solid-state polymorphic transition of glycine in mechano-chemical processing. *Chem. Eng. Res. Des.* **88**, 1169–1173 (2010).
25. Rezaee, M. & Mousavi Khoie, S. M. Mechanically induced polymorphic phase transformation in nanocrystalline TiO₂ powder. *J. Alloy Compd* **507**, 484–488 (2010).
26. Napolitano, E., Mulas, G., Enzo, S. & Delogu, F. Kinetics of mechanically induced anatase-to-rutile phase transformations under inelastic impact conditions. *Acta Mater.* **58**, 3798–3804 (2010).
27. Delogu, F., Mulas, G., Schiffrini, L. & Cocco, G. Mechanical work and conversion degree in mechanically induced processes. *Mat. Sci. Eng. A* **382**, 280–287 (2004).
28. Delogu, F., Orrù, R. & Cao, G. A novel macrokinetic approach for mechanochemical reactions. *Chem. Eng. Sci.* **58**, 815–821 (2003).
29. Santhanam, P. R. & Dreizin, E. L. Predicting conditions for scaled-up manufacturing of materials prepared by ball milling. *Powder Technol.* **221**, 403–411 (2012).
30. Rosenkranz, S., Breitung-Faes, S. & Kwade, A. Experimental investigations and modelling of the ball motion in planetary ball mills. *Powder Technol.* **212**, 224–230 (2011).
31. Mio, H., Kano, J. & Saito, F. Scale-up method of planetary ball mill. *Chem. Eng. Sci.* **59**, 5909–5916 (2004).
32. Broghini, M. et al. Modeling of the planetary ball-milling process: the case study of ceramic powders. *J. Eur. Ceram. Soc.* **36**, 2205–2212 (2016).
33. Cuadrado-Laborde, C., Damonte, L. C. & Mendoza-Zelis, L. Ball milling sound statistical analysis and its relationship with the mechanical work intensity. *Rev. Sci. Instrum.* **75**, 2334 (2004).
34. Le Brun, P., Froyen, L. & Delaey, L. The modelling of the mechanical alloying process in a planetary ball mill: comparison between theory and in-situ observations. *Mater. Sci. Eng. A* **161**, 75–82 (1993).
35. Gusev, V. G., Sobol'kov, A. V., Aborkin, A. V. & Alymov, M. I. Simulation of the energy-force parameters of planetary ball mill processing and estimation of their influence on the particle size in an AMg2 alloy/graphite composite powder. *Russ. Metall.* **2019**, 24–30 (2019).
36. Rogachev, A. S. et al. Experimental investigation of milling regimes in planetary ball mill and their influence on structure and reactivity of gasless powder exothermic mixtures. *Powder Technol.* **274**, 44–52 (2019).
37. Schilz, J. Internal Kinematics of Tumbler and Planetary Ball Mills: a mathematical model for the parameter setting. *Mater. Trans.* **39**, 1152–1157 (1998).
38. Cahn, J. W. The kinetics of grain boundary nucleated reactions. *Acta Metall.* **4**, 449–459 (1956).
39. Zaug, J. M., Soper, A. K. & Clark, S. M. Pressure-dependent structures of amorphous red phosphorus and the origin of the first sharp diffraction peaks. *Nat. Mater.* **7**, 890–899 (2008).
40. Ruck, M. et al. Fibrous red phosphorus. *Angew. Chem. Int. Ed. Engl.* **44**, 7616–7619 (2005).
41. Rissi, E. N., Soignard, E., McKiernan, K. A., Benmore, C. J. & Yarger, J. L. Pressure-induced crystallization of amorphous red phosphorus. *Solid State Commun.* **152**, 390–394 (2012).
42. Bouffelfel, S. E., Seifert, G., Grin, Y. & Leoni, S. Squeezing lone pairs: the A17 to A7 pressure-induced phase transition in black phosphorus. *Phys. Rev. B* **85**, 014110 (2012).
43. Jacobs, R. B. Phosphorus at high temperatures and pressures. *J. Chem. Phys.* **5**, 945–953 (1937).
44. Xiang, H. et al. The mechanism of structural changes and crystallization kinetics of amorphous red phosphorus to black phosphorus under high pressure. *Chem. Commun.* **55**, 8094–8097 (2019).
45. Yang, Z. et al. Field-effect transistors based on amorphous black phosphorus ultrathin films by pulsed laser deposition. *Adv. Mater.* **27**, 3748–3754 (2015).
46. Smith, J. B., Hagaman, D. & Ji, H. F. Growth of 2D black phosphorus film from chemical vapor deposition. *Nanotechnology* **27**, 215602 (2016).
47. Suryanarayana, C. Mechanical alloying and milling. *Prog. Mater. Sci.* **46**, 1–184 (2001).
48. Huberty, J. & Xu, H. Kinetics study on phase transformation from titania polymorph brookite to rutile. *J. Solid State Chem.* **181**, 508–514 (2008).
49. Will, G. & Berndt, H. Kinetics of the pressure-induced first-order phase transformation of RbJ. *High Press. Res.* **21**, 215–225 (2001).
50. Ocenasek, J., Novak, P. & Agbo, S. Finite-thickness effect on crystallization kinetics in thin films and its adaptation in the Johnson-Mehl-Avrami-Kolmogorov model. *J. Appl. Phys.* **115**, 043505 (2014).
51. Sun, N. X., Liu, X. D. & Lu, K. An explanation to the anomalous avrami exponent. *Scr. Mater.* **34**, 1201–1207 (1996).
52. Chandra Shekar, N. V. & Rajan, K. G. Kinetics of pressure induced structural phase transitions—a review. *Bull. Mater. Sci.* **24**, 1–21 (2001).
53. Elkin, F. S., Tsiok, O. B., Khvostantsev, L. G. & Brazhkin, V. V. Precise in situ study of the kinetics of pressure-induced phase transition in CaF₂ including initial transformation stages. *J. Exp. Theor. Phys.* **100**, 971–976 (2005).
54. Ye, F. & Lu, K. Crystallization kinetics of amorphous solids under pressure. *Phys. Rev. B* **60**, 7018–7024 (1999).
55. Li, X. et al. Synthesis of thin-film black phosphorus on a flexible substrate. *2D Materials* **2**, 031002 (2015).
56. Bridgman, P. W. Theoretically interesting aspects of high pressure phenomena. *Rev. Mod. Phys.* **7**, 1–33 (1935).
57. Keyes, R. W. The electrical properties of black phosphorus. *Phys. Rev.* **92**, 580–584 (1953).
58. Shoichi, E., Yuichi, A., Shin-ichi, T. & Shin-ichiro, N. Growth of large single crystals of black phosphorus under high pressure. *Jpn J. Appl. Phys.* **21**, L482–L484 (1982).
59. Bridgman, P. W. Shearing phenomena at high pressures, particularly in inorganic compounds. *Proc. Am. Acad. Arts Sci.* **71**, 387–460 (1937).
60. Zadorozhnyy, V. Y., Shelekhov, E. V., Milovzorov, G. S., Strugova, D. V. & Zinnurova, L. K. Analysis of the background temperature during the mechanical alloying of metal powders in the planetary ball mill. *Inorg. Mater. Appl. Res.* **9**, 559–565 (2018).
61. Andersen, J. M. & Mack, J. Decoupling the Arrhenius equation via mechanochemistry. *Chem. Sci.* **8**, 5447–5453 (2017).
62. Schmidt, R., Martin Scholze, H. & Stolle, A. Temperature progression in a mixer ball mill. *Int. J. Ind. Chem.* **7**, 181–186 (2016).
63. Takacs, L. & McHenry, J. S. Temperature of the milling balls in shaker and planetary mills. *J. Mater. Sci.* **41**, 5246–5249 (2006).
64. Delogu, F. & Cocco, G. Weakness of the “hot spots” approach to the kinetics of mechanically induced phase transformations. *J. Alloy Compd* **465**, 540–546 (2008).
65. Delogu, F. A few details of the austenite to martensite phase transformation in 304 stainless steel powders under mechanical processing. *Acta Mater.* **59**, 2069–2074 (2011).
66. Levitas, V. I. High pressure phase transformations revisited. *J. Phys. Condens. Matter* **30**, 163001 (2018).

67. Ji, C. et al. Shear-induced phase transition of nanocrystalline hexagonal boron nitride to wurtzitic structure at room temperature and lower pressure. *Proc. Natl. Acad. Sci. USA* **109**, 19108–19112 (2012).
68. Gasgnier, M., Szwarc, H. & Ronez, A. Low-energy ball-milling: transformations of boron nitride powders. Crystallographic and chemical characterizations. *J. Mater. Sci.* **35**, 3003–3009 (2000).
69. Levitas, V. I. & Javanbakht, M. Phase transformations in nanograin materials under high pressure and plastic shear: nanoscale mechanisms. *Nanoscale* **6**, 162–166 (2014).
70. Gilman, P. S. & Benjamin, J. S. Mechanical alloying. *Ann. Rev. Mater. Sci.* **13**, 279–300 (1983).
71. Roth, W. L., DeWitt, T. W. & Smith, A. J. Polymorphism of red phosphorus. *J. Am. Chem. Soc.* **69**, 2881–2885 (1947).
72. Jiang, H., Shi, H., Sun, X. & Gao, B. Optical anisotropy of few-layer black phosphorus visualized by scanning polarization modulation microscopy. *ACS Photon.* **5**, 2509–2515 (2018).

ACKNOWLEDGEMENTS

This work has been funded, in part, by Boise State University College of Engineering Seed Funding Program and Idaho Commerce IGEM (grant number 003786). Materials were synthesized and processed in the Advanced Materials Laboratory (Boise State University, Boise, ID). XRD, SEM, and TEM were performed in the Boise State Center for Materials Characterization (Boise State University, Boise, ID). Raman spectroscopy was performed in the Advanced Nano Materials and Manufacturing Laboratory (Boise State University, Boise, ID). Photoluminescence spectroscopy was performed at the Center for Nanoscale Materials, an Office of Science user facility which is supported by the U.S. Department of Energy, Office of Science, Office of Basic Energy Sciences, under Contract No. DE-AC02-06CH11357 as well as by the U.S. Department of Energy's Energy Efficiency & Renewable Energy (EERE) Advance Manufacturing Office (AMO) through the Argonne Chain Reaction Innovations Program through CRADA No. A18169 with Iris Light Technologies, Inc.

AUTHOR CONTRIBUTIONS

S.V.P. conceptualized and executed the milling studies, performed and analyzed the characterization data. C.H. performed PL measurements. F.M. and J.D.W. discussed the data and provided useful suggestions. S.V.P. wrote the manuscript with

comments and edits from all co-authors. D.E. and B.J.J. supervised and directed the project.

COMPETING INTERESTS

The authors declare no competing interests.

ADDITIONAL INFORMATION

Supplementary information is available for this paper at <https://doi.org/10.1038/s41699-020-00170-4>.

Correspondence and requests for materials should be addressed to B.J.J.

Reprints and permission information is available at <http://www.nature.com/reprints>

Publisher's note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2020