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Determining Recharge Through Small-Scale Geochemical Analysis on Samaria Mountain Springs

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Abstract

The Samaria Mountains in Southeastern Idaho are a crucial water recharge area for the Malad Valley and the Lower Malad River. The study area is managed by the Bureau of Land Management and is used for recreation and cattle grazing. Springs on the cattle allotment used for troughs and water storage have not yet been included in Malad Valley hydrology studies. To address sustainable use of these springs, we are providing spring systems recharge assessment. We have created a hydrochemical profile for the Samaria Mountains for comparison with a similar well-studied research site in southwestern Idaho. Through USGS standards, streams samples were collected and tested for cations (Ca, Mg, Na, K), anions (Br, nitrate, nitrite, sulfate, phosphate, chloride, and fluoride), and deuterium and oxygen-18 isotopes. The ion concentrations and pH recordings from the spring samples reflect the underlying geology of carbonate rocks. However, the isotope ratios are similar to Dry Creek Experimental Watershed ratios, where snowmelt is the primary contributor to groundwater. Accordingly, recharge for the Samaria Mountain spring system is assessed as sourced from snowpack and calls for quantification of average annual snowpack contribution to groundwater.

Introduction:

The Samaria Mountains southwest of Malad, Idaho are considered a recharge area for the the Malad River Basin (Hurlow, 2016). this area has been continually excluded from studies of the Malad River Basin. (Pluhowski 1970, Hurlow 2016).

Over the course of the last three years, the Bureau of Land Management State Office Water Rights Department has been surveying water sources on BLM land within the Bear River Adjudication. Measuring springs and stream consistency is crucial for BLM management practices (Barquin 2008). Understanding the complex recharge rates for springs on BLM land assists in determining the reliance of that stream throughout the year and dry seasons.

Research Goal:

The purpose of this study is to create a hydrochemical profile to assist recharge analysis for the spring systems in the Samaria Mountains.

Geologic Context:

The Samaria Mountains are a part of a western dipping ramp of carbonate rocks referred to as the Malad Ramp (Rodgers et al. 1992). Most of the Permian limestone and dolomite are western dipping at 5-20 degrees. Similar units are seen across the valley in the Portneuf range bounded by a thrust fault (Rodgers et al. 1992, Figure 1). In this specific area, there are multiple contacts between carbonate rocks. The presence of normal faulting and uplifted limestones could possibly produce a large recharge area for springs in the Samaria Mountains and other areas along the Malad Ramp.

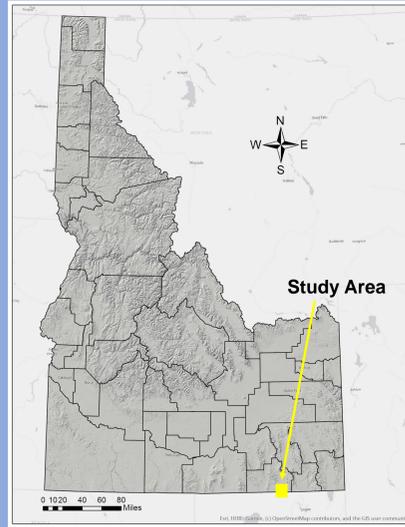


Figure 1: Map of Idaho with outlined counties. Area for this study is marked with a yellow box. Created in ArcGIS Pro.

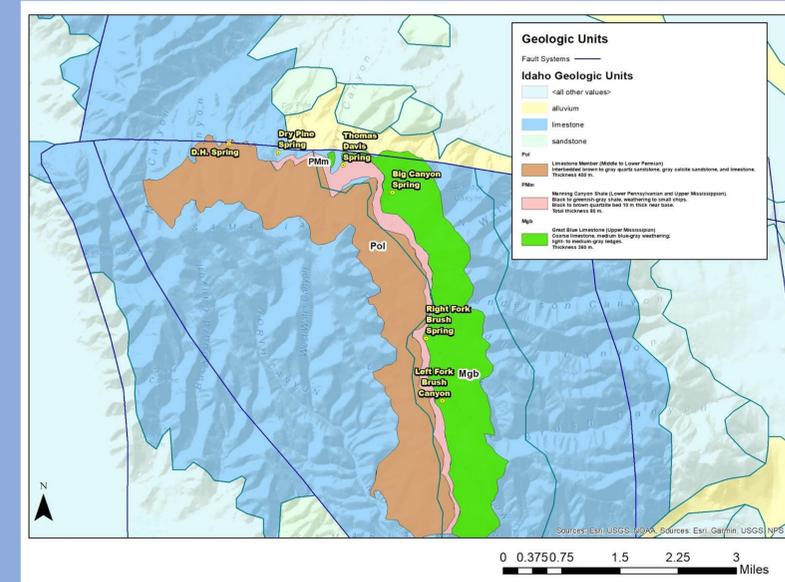


Figure 2: Geologic Map of the Samaria Mountains. Spring locations are marked with a yellow dot and name. Geologic units. General Idaho Geologic map units were sourced from Stoesser et al. (2005). The individual rock facies were created in ArcMap by outlining units from Platt (1977). Figure created in ArcMap.

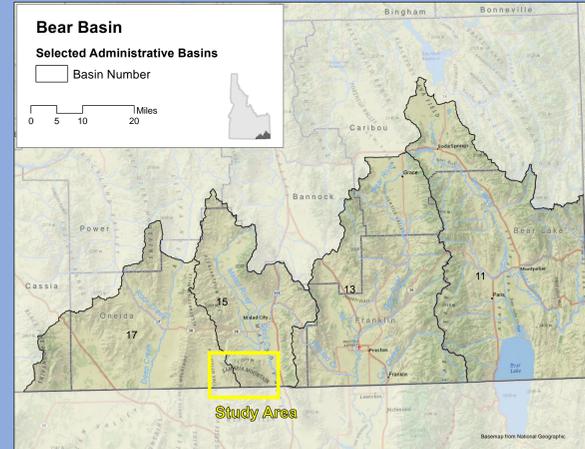


Figure 3: Map of the Bear River Basin Adjudication. Study area is outlined with a yellow square. Study area took place in Basin 15. Modified from Idaho Department of Water Resources - <https://idwr.idaho.gov/water-rights/adjudication/BRBA/>

Methods:

Spring sites were surveyed in Hydrologic Basin 15 (Fig 3). Samples were collected at the start of the water year (early October), due to the low amounts of predicted rainfall in the Malad River Valley (Pluhowski 1970). On site data collection process was created on the basis of USGS standards (Hem 1989).

- On site data included pH, conductivity, and temperature at each stream location.
- Anion analysis was conducted for bromide, nitrates, nitrites, sulfate, chloride, phosphate, and fluoride on a Lachat QuikChem8500. Samples were run in duplicate, and reproducibility was evaluated through internal laboratory standards.
- HCO₃⁻ and CO₃⁻² were calculated through testing for alkalinity with a Hach Alkalinity Test Kit. Anion concentrations were processed through MINTEQ.
- Cation concentrations (Ca, Mg, Na, K) were collected by Analytical Laboratories in Boise, ID, and measured through the EMA Method 200.7.
- The six stream samples were analyzed for δ²H and δ¹⁸O through a Los Gatos Research Liquid Water Isotope Analyzer (LWIA).
- Homogeneity of spring sources will be assessed through a Piper plot. δ²H/δ¹⁸O ratios will be plotted against the Local Meteoric Water Line (LMWL) for Southeastern Idaho, Montana, and Wyoming (Benjamin et al. 2005). Isotope results from study area will be compared to a watershed under similar climate and time frame. Comparison to the Dry Creek Experimental Watershed will be employed (Tetzlaff et al. 2015)

Table 1: On-Site Collected Data, 10-12-2019

Site	Abr.	Conduct. (us/m)	Temp (C)	pH	Temp (C)
Left Fork	LB	0.453	8.6	7.451	7.3
Right Fork	RB	0.664	8.6	7.982	7.2
D.H.	DH	0.461	7.9	8.14	6.9
Dry Pine	DP	0.444	8.4	8.076	6.9
Big	Big	0.482	8.2	8.408	7.6
Thomas	TD	0.482	8.2	8.408	7.6

Results:

Cation/Anion Analysis

- High concentrations of Ca and HCO₃ and CO₃. Low amounts of Cl milliequivalents
- No visible correlation between stream sites and flow path (Fig. 2, Fig. 5)

Isotope Analysis

- δ¹⁸O distribution is 1.526 ‰, and δ²H range is 10.73 ‰. This small spread is not large enough to delineate a trend within the data.
- All samples had neutral pH: range from 7.45 - 8.408.

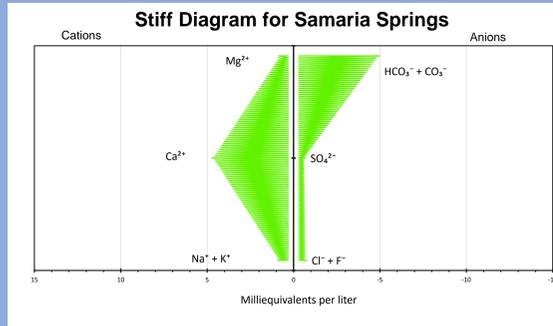


Figure 4: Stiff diagram created from the average ion milliequivalents of all 6 spring sites outlined in Table 1. Graph created through Halford Hydrology LLC Piper and Stiff diagram Excel spreadsheet.

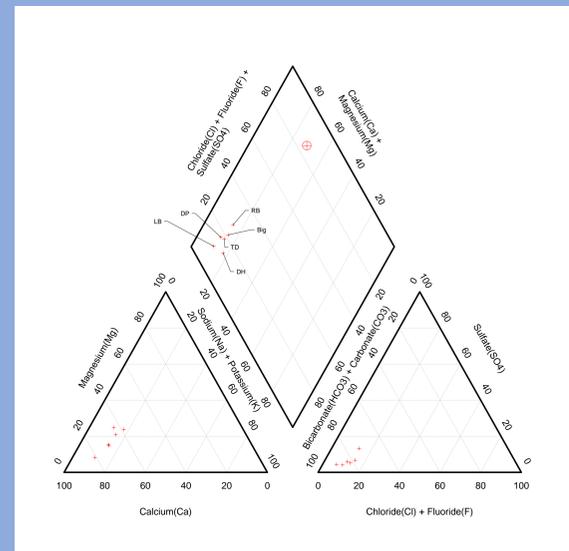


Figure 5: Piper Plot of Cation and anion milliequivalents for each spring. Each spring is labeled by their abbreviation (see Table 1). Standards for plotting ion data is outlined by Hem (1989). Piper plot created through a USGS Nevada Microsoft Excel template. - <https://nevada.usgs.gov/tech/excel/hydrology/WaterQualityTools>

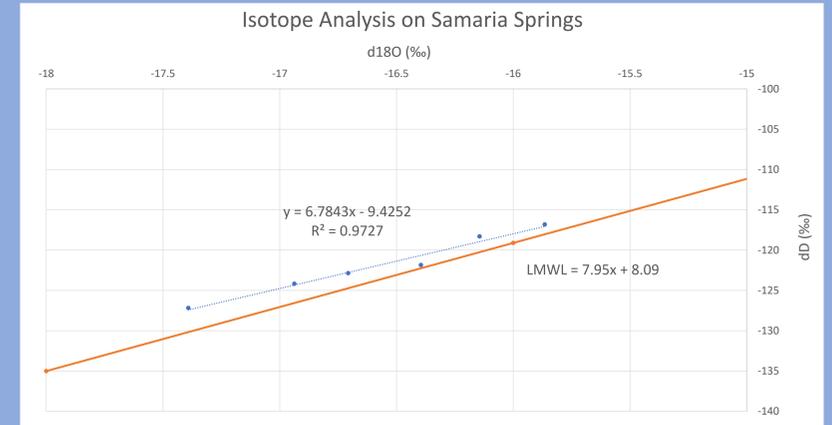


Figure 6: Graph of δ²H vs δ¹⁸O ratios for Samaria Spring sites. Measurements were collected through a LWIA at Boise State University. Reproducibility through internal laboratory standards. A linear regression line was created through Microsoft Excel. Spring stable isotope measurements are plotted along side a LMWL for Southeastern Idaho (Benjamin et al. 2005).

Discussion

- Groundwater sourced from carbonate rocks consists mostly of Ca-Mg-HCO₃ over Cl and sulphate and typically have a pH of over 7 due to the alkalinity of the ions (Singhal and Gupta 2010). All samples collected from the Samaria mountains follow this criteria.
- Figure 4 contradicts Stiff diagram for surface water measurements downstream of spring systems (Pluhowski 1970).
- Range of stable isotope values are close to those measured in the Dry Creek Watershed (Tetzlaff et al. 2015). Snowmelt is the main contributor to groundwater flow and recharge. Samples taken during snow runoff are needed for further analysis.
- All values plot above the LMWL, which could indicate a more localized MWL for the area. But, more isotope analysis is needed for the area to determine if this is a significant trend. Year round isotope data is needed.
- Stable isotope data most likely reflects localized MWL.

Conclusions

- The ion concentrations for the Samaria Mountains reflect the underlying geology in the study area, and therefore the water recharging the springs is groundwater. More studies on groundwater flow are need to confirm the relationship between flow and the physical rock units. No flow path can be determined from the high altitude springs. More samples at interval distances are needed for a clearer analysis.
- δ²H/δ¹⁸O ratios have little variability in determining recharge source, but if we assume that conditions are similar throughout high altitude arid areas in Idaho, then recharge is primarily sourced from spring snowmelt. Year-round samples would assist in comparing δ²H/δ¹⁸O ratios to spring runoff measurements.

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